PATENT COOPERATION TREATY

From the INTERNATIONAL BUREAU

PCT

NOTIFICATION OF RECEIPT OF RECORD COPY

(PCT Rule 24.2(a))

BRASS, Daniel Dizengoff Street 10 64281 Tel Aviv ISRAËL

Date of mailing (day/month/year) 07 August 2000 (07.08.00)	IMPORTANT NOTIFICATION
Applicant's or agent's file reference D01/5	International application No. PCT/IL00/00364

The applicant is hereby notified that the International Bureau has received the record copy of the international application as detailed below.

Name(s) of the applicant(s) and State(s) for which they are applicants:

DEXCEL LTD. (for all designated States except US)

LAHAV, Raffael et al (for US)

International filing date

21 June 2000 (21.06.00)

Priority date(s) claimed

22 June 1999 (22.06.99)

Date of receipt of the record copy by the International Bureau

11 July 2000 (11.07.00)

List of designated Offices

AP:GH,GM,KE,LS,MW,MZ,SD,SL,SZ,TZ,UG,ZW

EA: AM, AZ, BY, KG, KZ, MD, RU, TJ, TM

EP:AT,BE,CH,CY,DE,DK,ES,FI,FR,GB,GR,IE,IT,LU,MC,NL,PT,SE

OA:BF,BJ,CF,CG,CI,CM,GA,GN,GW,ML,MR,NE,SN,TD,TG

National :AE,AG,AL,AM,AT,AU,AZ,BA,BB,BG,BR,BY,BZ,CA,CH,CN,CR,CU,CZ,DE,DK,DM,DZ,EE,

ES,FI,GB,GD,GE,GH,GM,HR,HU,ID,IL,IN,IS,JP,KE,KG,KP,KR,KZ,LC,LK,LR,LS,LT,LU,LV,MA,

MD,MG,MK,MN,MW,MX,MZ,NO,NZ,PL,PT,RO,RU,SD,SE,SG,SI,SK,SL,TJ,TM,TR,TT,TZ,UA,UG,US,

UZ,VN,YU,ZA,ZW

ATTENTION

The applicant should carefully check the data appearing in this Notification. In case of any discrepancy between these data and the indications in the international application, the applicant should immediately inform the International Bureau.

In addition, the applicant's attention is drawn to the information contained in the Annex, relating to:

X time limits for entry into the national phase confirmation of precautionary designations

requirements regarding priority documents

A copy of this Notification is being sent to the receiving Office and to the International Searching Authority.

The International Bureau of WIPO 34, chemin des Colombettes 1211 Geneva 20, Switzerland Authorized officer:

Marie-José Devillard

Telephone No. (41-22) 338.83.38

Facsimile No. (41-22) 740.14.35

DAV

PATENT COOPERATION TREATY

From the INTERNATIONAL BUREAU

PCT

NOTIFICATION OF ELECTION

(PCT Rule 61.2)

Commissioner
US Department of Commerce
United States Patent and Trademark
Office, PCT
2011 South Clark Place Room
CP2/5C24
Arlington, VA 22202

Date of mailing (day/month/year) 19 March 2001 (19.03.01)	ETATS-UNIS D'AMERIQUE in its capacity as elected Office	
International application No. PCT/IL00/00364	Applicant's or agent's file reference D01/5	
International filing date (day/month/year) 21 June 2000 (21.06.00)	Priority date (day/month/year) 22 June 1999 (22.06.99)	
Applicant		
LAHAV, Raffael et al		

1.	The designated Office is hereby notified of its election made:
	X in the demand filed with the International Preliminary Examining Authority on:
	18 December 2000 (18.12.00)
	in a notice effecting later election filed with the International Bureau on:
2.	The election X was
	was not
	made before the expiration of 19 months from the priority date or, where Rule 32 applies, within the time limit under Rule 32.2(b).

The International Bureau of WIPO 34, chemin des Colombettes 1211 Geneva 20, Switzerland

Authorized officer

Claudio Borton

Telephone No.: (41-22) 338.83.38

Facsimile No.: (41-22) 740.14.35

PATENT COOPERATION TRUITY

	From the INTERNATIONAL BUREAU			
PCT	То:			
NOTIFICATION OF THE RECORDING OF A CHANGE (PCT Rule 92bis.1 and Administrative Instructions, Section 422) Date of mailing (day/month/year)	BRASS, Daniel Plinner, Bodner, Brass Beit Agish Ravad 13 Noach Mozes Street Tel Aviv 67442 ISRAËL			
19 March 2001 (19.03.01)				
Applicant's or agent's file reference D01/5	IMPORTANT NOTIFICATION			
International application No. PCT/IL00/00364	International filing date (day/month/year) 21 June 2000 (21.06.00)			
The following indications appeared on record concerning: the applicant				
Name and Address BRASS, Daniel Dizengoff Street 10 64281 Tel Aviv	State of Nationality State of Residence Telephone No.			
64281 ⁻ Tel Aviv Israel	972-3-620-1866 Facsimile No. 972-3-620-1469			
	Teleprinter No.			
The International Bureau hereby notifies the applicant that the the person X the name the add	[]			
Name and Address	State of Nationality State of Residence			
BRASS, Daniel Plinner, Bodner, Brass	Telephone No.			
Beit Agish Ravad 13 Noach Mozes Street	972-3-696-9090			
Tel Aviv 67442 Israel	Facsimile No.			
191001	972-3-696-6656			
	Teleprinter No.			
3. Further observations, if necessary:				
4. A copy of this notification has been sent to:				
X the receiving Office	the designated Offices concerned			
the International Searching Authority	X the elected Offices concerned			
X the International Preliminary Examining Authority	other:			
	Authorized officer			
The International Bureau of WIPO 34, chemin des Colombettes 1211 Geneva 20, Switzerland	Claudio Borton			
Faccimile No : (41-22) 740 14 35	Telephone No.: (41-22) 338.83.38			

PATENT COOPERATION TIL ATY

PCT

NOTIFICATION CONCERNING AMENDMENTS OF THE CLAIMS

(PCT Rule 62 and Administrative Instructions, Section 417)

Date of mailing (day/month/year) 19 March 2001 (19.03.01)

International application No. PCT/IL00/00364

Applicant

DEXCEL LTD. et al

From the INTERNATIONAL BUREAU

To:

Commissioner
US Department of Commerce
United States Patent and Trademark
Office, PCT
2011 South Clark Place Room
CP2/5C24
Arlington, VA 22202
ETATS-UNIS D'AMERIQUE

in its capacity as International Preliminary Examining Authority

International filing date (day/month/year)
21 June 2000 (21.06.00)

The International Bureau hereby informs the International Preliminary Examining Authority that no amendments under Article 19 have been received by the International Bureau (Administrative Instructions, Section 417).

The International Bureau of WIPO 34, chemin des Colombettes 1211 Geneva 20, Switzerland

Authorized officer

Claudio Borton

Telephone No. (41-22) 338.83.38

Facsimile No. (41-22) 740.14.35

003905757

PATENT COOPERATION TRUITY

	From the INTERNATIONAL BUREAU			
PCT	То:			
NOTIFICATION OF THE RECORDING OF A CHANGE (PCT Rule 92bis.1 and Administrative Instructions, Section 422) Date of mailing (day/month/year) 19 March 2001 (19.03.01)	BRASS, Daniel Plinner, Bodner, Brass Beit Agish Ravad 13 Noach Mozes Street Tel Aviv 67442 ISRAËL			
Applicant's or agent's file reference	IMPORTANT NOTIFICATION			
D01/5	INITOTITATI NOTIFICATION			
International application No. PCT/IL00/00364	International filing date (day/month/year) 21 June 2000 (21.06.00)			
The following indications appeared on record concerning: the applicant the inventor	the agent the common representative			
Name and Address BRASS, Daniel	State of Nationality State of Residence			
Dizengoff Street 10 64281 Tel Aviv Israel	Telephone No. 972-3-620-1866			
	Facsimile No. 972-3-620-1469			
	Teleprinter No.			
2. The International Bureau hereby notifies the applicant that the the person X the name the add				
Name and Address	State of Nationality State of Residence			
BRASS, Daniel Plinner, Bodner, Brass Beit Agish Ravad	Telephone No. 972-3-696-9090			
13 Noach Mozes Street Tel Aviv 67442 Israel	Facsimile No.			
	972-3-696-6656 Teleprinter No.			
3. Further observations, if necessary:				
3. Further observations, if necessary:				
4. A copy of this notification has been sent to:				
X the receiving Office	the designated Offices concerned			
the International Searching Authority	X the elected Offices concerned			
X the International Preliminary Examining Authority	other:			
The International Bureau of WIPO 34, chemin des Colombettes 1211 Geneva 20, Switzerland	Authorized officer Claudio Borton			
Facsimile No.: (41-22) 740.14.35	Telephone No.: (41-22) 338.83.38			

PATENT COOPERATION TREATY

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REC'D 3 0 OCT 2001

INTERNATIONAL PRELIMINARY EXAMINATION REPORT

PCT

(PCT Article 36 and Rule 70)

14

Applicant's or agent's file reference		See Need			
D01/5	FOR FURTHER ACTION	Preliminary Examination Report (Form			
International application No.	International filing date (day/	month/year)	Priority date (day/month/year)		
PCT/IL00/00364 21 JUNE 2000 22 JUNE 1999			22 JUNE 1999		
International Patent Classification (IPC) or national classification and IPC Please See Supplemental Sheet.					
Applicant DEXCEL LTD.					
 This international preliminary examination report has been prepared by this International Preliminary Examining Authority and is transmitted to the applicant according to Article 36. This REPORT consists of a total of sheets. This report is also accompanied by ANNEXES, i.e., sheets of the description, claims and/or drawings which have been amended and are the basis for this report and/or sheets containing rectifications made before this Authority. (see Rule 70.16 and Section 607 of the Administrative Instructions under the PCI). 					
These annexes consist of a tot	$\boldsymbol{\wedge}$		401 410 101).		
3. This report contains indication		ems:			
T Desir selle					
<u> </u>					
II Priority					
	t of report with regard to no	velty, inventi	ve step or industrial applicability		
IV Lack of unity of i	invention				
V X Reasoned statement citations and explan	under Article 35(2) with regalations supporting such statemen	rd to novelty, ent	inventive step or industrial applicability,		
VI Certain documents o	ited				
VII Certain defects in the international application					
VIII Certain observations on the international application					
Date of submission of the demand Date of completion of this report			of this report		
18 DECEMBER 2000	18 DECEMBER 2000 17 SEPTEMBER 2001				
Name and mailing address of the IPEA/	Name and mailing address of the IPEA/US Authorized officer				
Commissioner of Patents and Trademarks Box PCT					
Washington, D.C. 20231					
Facsimile No. (703) 305-3230 Telephone No. (703) 308-0196			03) 308-0196		

INTERNATIONAL PRELIMINARY EXAMINATION REPORT

International	application	No
imernational	application	NO.

PCT/IL00/00364

I. B	asis of	the report		
1. Witl	h regard	to the elements of the interna	ational application *	
x		iternational application as	* -	•
x	l	escription:		
		1-21		as originally filed
		NONE		
			, filed with the letter of	
X		aims: 22-26		
				, as originally filed
		NONE	, as amended (together with a	
			, filed with the letter of	, med with the demand
			, , , , , , , , , , , , , , , , , , , ,	
\mathbf{x}	the dr	rawings:		
	pages	NONE		, as originally filed
		NONE		, filed with the demand
	pages	NONE	, filed with the letter of	
r.	41			
X		quence listing part of the d	-	
				, as originally filed
	nages	NONE	, filed with the letter of	, filed with the demand
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The	internat se elem the lar the lar	ional application was filed, the ents were available or furnishing uage of a translation furnishing uage of publication of the guage of the translation furnishing the standard of the translation furnishing uage of the translation furnishing uage.	tents marked above were available or furnished to this inless otherwise indicated under this item. The to this Authority in the following language rnished for the purposes of international search the international application (under Rule 48.30) is the formula of the purposes of international preliminary in the purpose in the p	which is: th (under Rule 23.1(b)).
3. Wit	h regar liminar	d to any nucleotide and/o y examination was carried	ramino acid sequence disclosed in the internation out on the basis of the sequence listing:	onal application, the international
لـــا	contain	ned in the international a	oplication in printed form.	
	filed to	ogether with the internation	onal application in computer readable form.	·
	furnish	ned subsequently to this A	authority in written form.	
=			authority in computer readable form.	
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	interna	tional application as filed l	tly furnished written sequence listing does not go has been furnished.	beyond the disclosure in the
	The sta		recorded in computer readable form is identical to	the writen sequence listing has
4 X	The ar	nendments have resulted	in the cancellation of	
4.	T-			
ļ		he description, pages	NONE	
	╠╣╵	he claims, Nos.	NONE	
	X t	he drawings, sheets /fig _	NONE	
5.	This re	port has been drawn as if (so	ome of) the amendments had not been made, since t	hev have been considered to go
	beyon	d the disclosure as filed, as in	ndicated in the Supplemental Box (Rule 70.2(c)).**	_
and 7	3 7 epor 70.17).	i as originally filea and a	hed to the receiving Office in response to an invitation are not annexed to this report since they do not con	ntain amendments (Rules 70.16
**Any	replace	ment sheet containing such	amendments must be referred to under item 1 and	annexed to this report.

INTERNATIONAL PRELIMINARY EXAMINATION REPORT

International application No.

PCT/IL00/00864

1. statement			
statement			
Novelty (N)	Claims	NONE	YE
	Claims	1-39	NO
Inventive Step (IS)	Claims	NONE	VII
	Claims	1-39	YE
Industrial Applicability (IA)	Claims	1-39	YE
	Claims	NONE	NO
citations and explanations (Rule			
Claims 1-39 lack novelty under PCT Article terein after as the '338 patent.	33(2) as being a	iticipated by Bergstrand et al. (US Paten)	: No. 5,817, 338) referred
The '338 patent discloses a pharmaceutical			
opolymer, triethyl citrate, mono- and di-gloomprises a core of active omeprazole, a sep The omeprazole tablet formulation is also fo	arating layer, an	enteric coating pellets and material and	tableting composition
nd claims 1-25. The '338 patent thus anti Laims 1-39 meet the criteria set out in PC erivative such as omeprazole composition formulation industry.	icipates the clair T Article 33(4), ormulated into :	ned invention. because the instant invention directed to	o coated benzimidazole
nd claims 1-25. The '338 patent thus anti Claims 1-39 meet the criteria set out in PC erivative such as omeprazole composition fo	icipates the clair T Article 33(4), ormulated into :	ned invention. because the instant invention directed to	o coated benzimidazole
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INTERNATIONAL PRELIMINARY EXAMINATION REPORT

International application No.

PCT/IL00/00364

Supp	lemental	Box
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(To be used when the space in any of the preceding boxes is not sufficient)

Continuation of: Boxes I - VIII

Sheet 10

CLASSIFICATION:

The International Patent Classification (IPC) and/or the National classification are as listed below:

IPC(7): A61K 9/00, 9/14, 9/16, 9/20, 9/22, 9/28, 9/50 and US Cl.: 424/400, 464, 465, 468, 474, 489, 490, 494, 495

INTERNATIONAL SEARCH REPORT

International application No. PCT/IL00/00364

A. CLAS	SSIFICATION OF SUBJECT MATTER			
• • •				
US CL:424/400, 464, 465, 468, 474, 489, 490, 494, 495 According to International Patent Classification (IPC) or to both national classification and IPC				
B. FIELDS SEARCHED				
	ocumentation searched (classification system followed	hy classification symbols)		
	•	by classification symbols,		
0.5. :	424/400, 464, 465, 468, 474, 489, 490, 494, 495			
Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched				
Electronic d	Electronic data base consulted during the international search (name of data base and, where practicable, search terms used)			
C. DOC	UMENTS CONSIDERED TO BE RELEVANT			
Category*	Citation of document, with indication, where ap	propriate, of the relevant passages	Relevant to claim No.	
X,P	US 6,013,281 A (LUNDBERG et al.) 11 January 2000, see abstract, column 5, line 6 to column 6 and line 49, column 7, line 27 to column 8, and line 24, examples 1-7 and claims 1-6 and 15-21.			
X	US 5,817,338 A (BERGSTRAND et al.) 06 October 1998, see entire document.			
X	US 5,753,265 A (BERGSTRAND et al.) 19 May 1998, see columns 5 and 6, column 7 line 43 to column 10, and line 48, examples 1-16 and claims 1-11, 14, 15, 17 and 19.			
x	US 4,853,230 A (LOVGREN et al.) 01 August 1989, see entire document.		1-39	
Furt	her documents are listed in the continuation of Box C	See patent family annex.		
	* Special categories of cited documents: *T* later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention			
to	be of particular relevance	"X" document of particular relevance; the		
	rlier document published on or after the international filing date	considered novel or cannot be considered when the document is taken alone	ered to involve an inventive step	
ci	cited to establish the publication date of another citation or other			
1	ecial reason (as specified)	considered to involve an inventive	e step when the document is	
	O document referring to an oral disclosure, use, exhibition or other combined with one or more other such documents, such combination being obvious to a person skilled in the art			
	ocument published prior to the international filing date but later than e priority date claimed	"&" document member of the same pater	nt family	
	actual completion of the international search	Date of mailing of the international se	earch report	
24 SEPT	24 SEPTEMBER 2000 2 4 OCT 2000			
Commissi	Name and mailing address of the ISA/US Commissioner of Patents and Trademarks Box PCT			
	Box PCT Washington, D.C. 20231 BLESSING FUBARA			
Facsimile 1	No. (703) 305-3230	Telephone No. (703) 308-8374		

Form PCT/ISA/210 (second sheet) (July 1998)*

10/018992 531 Rec'd PC 26 DEC 2001

APPLICATION FOR PATENT

TITLE:

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STABLE BENZIMIDAZOLE FORMULATION

INVENTORS:

RAFFAEL LAHAV and VALERIE AZOULAY

This Application is a Continuation-in-Part Application of PCT Application No. PCT/IL00/00364, filed on June 21, 2000, and also claims priority from Israeli Patent Application No. 130602, filed on June 22, 1999, both of which are hereby incorporated by reference as if fully set forth herein.

10 FIELD AND BACKGROUND OF THE INVENTION

The present invention relates to a novel stable formulation for an acid labile benzimidazole, and methods of preparation and administration thereof, and in particular, for a stable formulation of a benzimidazole which is suitable for oral administration.

Omeprazole, Pantoprazole, Lansoprazole and other derivatives of benzimidazole, which are active proton pump inhibitors and used conventionally for decreasing gastric secretion are known to be susceptible to degradation and transformation in acid media. Omeprazole, 5-methoxy-2(((4-methoxy-3,5-dimethyl-2-pyridinyl)methyl)sulfinyl)-1H-benzimidazole, is disclosed and described in European Patent No. 5129 and European Patent No. 124495, as well as in numerous other patents and published patent applications.

The susceptibility of these active proton pump inhibitor substances to degradation and transformation in acid media increases the difficulty of preparing a pharmaceutical form designed for oral administration. If the active substance comes into contact with the

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stomach content, which is a highly acidic medium, these chemical substances become degraded. Thus, these benzimidazole derivatives should be protected both during storage and during their passage through the acidic environment of the stomach.

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The stability of Omeprazole has been extensively studied (see for example A. Pilbrant and C. Cederberg, *Scan. J. Gastroenterol.*, **20**: 113-120, 1985). Omeprazole degrades with a half-life of less than 10 minutes in an environment with pH values below 4.0. At pH 6.5, the half life of Omeprazole is 18 hours and at pH 11 about 300 days. Therefore, the environment of Omeprazole should be kept at a sufficiently high pH value in order to maintain the stability of the compound, in a formulation which is suitable as a product for oral administration, for example by locating Omeprazole within a core which also contains alkaline constituents. This leads to an alkaline reaction aimed at improving stability of the active substance during manufacture thereof and during storage of the pharmaceutical formulation.

In addition, such a formulation must protect Omeprazole from the acidic environment of the stomach, since if Omeprazole is given orally without any protective coating, it will degrade in the acid environment of the stomach. European Patent No. 237,200 discloses one solution, which is to directly coat the solid core containing Omeprazole, or another benzimidazole derivative, with an enteric coating layer.

However, this apparent solution to the instability of Omeprazole caused further complications, in that the alkaline core containing Omeprazole was found to react with the enteric coating, thereby causing the enteric coating to degrade. A solution to these further complications is disclosed in United Kingdom Patent Application No. 2,189,698, in which Omeprazole is contained within a solid active core, which is coated first with a

subcoating layer and then with an enteric coating layer. The enteric coating layer protects the Omeprazole during the passage through the stomach, while the subcoating layer protects the enteric coating layer from reacting negatively with the alkaline core containing Omeprazole.

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The background art describes other attempts to provide formulations which are suitable for oral administration of acid-labile substances. For example, PCT Application No. WO 97/12581 discloses a composition adapted for oral administration containing Omeprazole which specifically does not include alkaline-reacting compounds. Instead, the composition features a core composed of a nuclei and Omeprazole compressed together, an intermediate layer and an enteric layer.

European Patent Application No. 519,144 discloses a formulation for Omeprazole, which features a neutral (sugar) core. Omeprazole is sprayed onto the sugar core, after which an intermediate coating layer and an enteric coating layer are sprayed onto the core.

PCT Application No. WO 98/00114 discloses a modification to other background art formulations for Omeprazole, in which the intermediate subcoating layer is partially neutralized with an alkaline compound. However, this modified formulation still features the subcoating layer, which is a disadvantage in that it complicates the manufacturing process and increases the expense and difficulty of manufacture. Thus, the formulation disclosed in PCT Application No. WO 98/00114, like those disclosed in European Patent Application No. 519,144 and other background art references, has the disadvantage of requiring the intermediate layer.

PCT Application No. WO 83/00435 discloses a solid dosage form, such as a capsule or tablet, containing a pharmacologically active agent coated with an anionic polymer, which is insoluble in gastric juice and in intestinal juice below pH 7. The preferred anionic polymer is a partly methyl esterified methacrylic acid polymer in which the ratio of free carboxylic groups to ester groups is about 1:2. In contrast to the present invention, Omeprazole is not disclosed as one of the active agents.

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French Application No. 2,692,146 discloses stable compositions of microgranules of gastro-protected Omeprazole. The composition features a center of Omeprazole diluted in mannitol. This center is coated with an intermediate layer featuring mannitol. An enteric coating is then added over this intermediate layer. PCT Application No. WO 97/12581 discloses a formulation in which an intermediate layer between the core and an enteric coating contains silicium dioxide.

PCT Application No. WO 96/37195 discloses a formulation which lacks a subcoating layer, but which features a core containing titanium dioxide. Both the core containing Omeprazole and the enteric coating layer placed on top of the core include titanium dioxide as an ingredient. Unfortunately, titanium dioxide is only able to mask the discoloration caused by the reaction between Omeprazole and the enteric coating layer, but cannot prevent such an undesirable reaction. Thus, the disclosed formulation does not prevent the undesirable reaction between the benzimidazole derivative and the enteric coating, which is known in the art.

German Patent Application No. 196 26 045 A1 discloses a method for stabilising Omeprazole by coating small tablets or pellets, containing large amounts of mannitol,

with a subcoating of Eudragit L. The subcoating of Eudragit L is neutralized, after which a final enteric coat of non-neutralized Eudragit L is applied.

A formulation of a benzimidazole derivative, such as Omeprazole, which lacks an intermediate coating layer and yet which is stable both during storage and during the passage through the stomach, would be highly desirable. Such a formulation would be simpler to manufacture and would expose the sensitive benzimidazole derivative to fewer production steps, thereby decreasing the possibility that the active compound would degrade during production. Unfortunately, such a stable benzimidazole formulation, which lacks an intermediate layer, is not currently available.

There is thus a unmet need for, and it would be useful to have, a stable benzimidazole formulation, particularly for Omeprazole which lacks an intermediate layer and yet which is stable both during storage and during the passage through the stomach.

SUMMARY OF THE INVENTION

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The formulation of the present invention contains a benzimidazole derivative, such as Omeprazole, and is able to maintain the stability of this active ingredient without a separating layer. Instead, the enteric coating layer is applied as a solution with a pH value of at least 6.5, and more preferably in a range of from about 7 to about 10, directly to the benzimidazole derivative substrate. This solution, with the optional addition of a plasticizer, can be directly coated onto the substrate without any necessity for an intermediate layer. Furthermore, in this pH range, the enteric coating is optionally

applicable in an aqueous solution, thereby obviating the need for organic solvents for dissolving the enteric coating material.

The resultant formulation maintains the stability of the benzimidazole derivative during storage and at the same time protects the product during passage through the acidic environment of the stomach, where the acidic environment of the stomach causes a partial ionic exchange to occur within the material of the coating. This partial ionic exchange renders the coating impermeable to the acidic liquids of the stomach. On the other hand, during storage the problem of interaction between the enteric coat and the alkaline core is thus completely eliminated as the "enteric coat" is no longer acidic during the storage period.

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Preferably, the benzimidazole derivative is selected from the group consisting of Omeprazole, Pantoprazole, Lansoprazole, Leminoprazole, Perprazole, Rabeprazole, and pharmaceutically acceptable salts thereof, as well as any other derivatives of benzimidazole which are proton pump inhibitors and which are conventionally used to decrease gastric secretion.

According to the present invention, there is provided a stable composition for a benzimidazole derivative, the composition comprising: (a) a substrate, the substrate featuring the benzimidazole derivative; and (b) an enteric coating material layered over the substrate, the enteric coating material having a pH value of at least about 6.5.

The substrate can optionally have several different structures. For example, the substrate is optionally an active core containing the benzimidazole derivative, in which the core is a pellet, bead or tablet for example. The active core can be prepared by any conventional method known in the art, including but not limited to, pellets prepared by

spheronisation, pellets prepared by coating an inert non pareil seed with Omeprazole, tablets prepared by granulation and compression, as well as any other methods.

The enteric coating material optionally and preferably includes an enteric material selected from the group consisting of hydroxypropyl methylcellulose phthalate, hydroxypropyl methylcellulose acetate succinate, polyvinyl acetate phthalate, cellulose acetate trimellitate, polymethacrylic acid methyl methacrylate and polymethacrylic acid ethyl methacrylate.

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More preferably, the enteric coating material further comprises an alkaline compound, such that the pH value is adjusted by adding the alkaline compound to the enteric material. Most preferably, the alkaline compound is an inorganic or organic alkaline salt compound. Even more preferably, the alkaline compound is selected from the group consisting of basic sodium, potassium or ammonium hydroxide. Also most preferably, the pH value is in a range of from about 7 to about 10.

The enteric coating material of the composition could optionally include a plasticizer. Preferably, the plasticizer is selected from the group consisting of a citric acid ester and a phthalic acid ester.

According to another embodiment of the present invention, there is provided a stable composition for a benzimidazole derivative, the composition consisting essentially of: (a) a substrate, the substrate featuring the benzimidazole derivative; and (b) an enteric coating material layered over the substrate, the enteric coating material having a pH value of at least about 6.5.

According to still another embodiment of the present invention, there is provided a method for producing a stable composition for a benzimidazole derivative, the method

comprising the steps of: (a) forming a substrate with the benzimidazole derivative; (b) preparing an enteric coating material having a pH value of at least about 6.5; and (c) layering the enteric coating material over the substrate.

DESCRIPTION OF THE PREFERRED EMBODIMENTS

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The formulation of the present invention contains a benzimidazole derivative, such as Omeprazole, and is able to maintain the stability of this active ingredient without a separating layer between the active compound and an enteric coating layer. Instead, the enteric coating layer is applied as a solution with a pH value of at least 6.5, and more preferably in a range of from about 7 to about 10, directly to the benzimidazole derivative substrate. This solution, with the addition of a plasticizer, can be directly coated onto the substrate without any necessity for an intermediate layer. Furthermore, in this pH range, the enteric coating is optionally applicable in an aqueous solution, thereby obviating the need for organic solvents for dissolving the enteric coating material.

The resultant formulation maintains the stability of the benzimidazole derivative during storage and at the same time protects the product during passage through the acidic environment of the stomach. The problem of interaction between the enteric coat and the alkaline core is thus completely eliminated as the enteric coat at this stage is no longer acidic.

Without wishing to be limited to a single mechanism, it is hypothesized that as the formulation passes through an acidic environment, such as the acidic environment of the stomach, the outer layer of the enteric coat is converted to an acidic form. This acidic form of the enteric coating material is insoluble in the acidic environment of the stomach.

If the formulation is then placed in an environment with a more alkaline pH value, for example by moving into the small intestine, the enteric coat dissolves and releases the active substance.

The use of an enteric coating which includes HPMCP

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(hydroxypropylmethylcellulose phthalate) neutralized with a basic salt is disclosed in U.S. Patent No. 5,225,202 and in two scientific articles, "Enteric Film Coating Using Completely Aqueous Dissolved Hydroxypropyl Methyl Cellulose Phthalate Spray Solutions" (J.W. Stafford et al., Drug Development and Industrial Pharmacy, 8:513-530, 1982) and "The In Vitro and In Vivo Performance of Aqueous Based Enteric Coats of Neutralized Hydroxypropyl Methyl Cellulose Phthalate" (J.R. Bloor et al., Drug Development and Industrial Pharmacy, 15:2227-2243, 1989). However, the disclosed enteric coating is not taught or suggested in any of these references as a suitable direct enteric coating for substrates which contain Omeprazole. As noted previously, Omeprazole and the related benzimidazole derivatives are unusually sensitive molecules, and as such must be carefully protected. Furthermore, U.S. Patent No. 5,225,202 teaches the necessity for a subcoat between the drug-containing substrate and the enteric coating for drugs which are not compatible with the enteric coating. By contrast, the present invention has been shown to be highly effective without such a subcoat, which is particularly surprising since the background art teaches that formulations containing Omeprazole or another benzimidazole derivative must also feature a subcoat. Neither scientific article even considers the problems associated with acid-sensitive drugs, and as such cannot teach or suggest the formulation of the present invention.

As shown by both the *in vitro* and *in vivo* data given below, the formulation of the present invention has been shown to be particularly effective for the oral administration of Omeprazole as the exemplary benzimidazole derivative, a result which could not have been predicted from these references. Indeed, the article by J.R. Bloor *et al.* teaches away from the use of such a neutralized enteric coating for any formulation, as this article disclosed good *in vitro* performance of the formulation but poor *in vivo* performance. By contrast, as described in greater detail below with regard to Example 7, the formulation of the present invention shows good performance *in vivo*. Thus, the background art neither teaches nor suggests the direct coating of a substrate containing Omeprazole or another benzimidazole derivative with an enteric coating material having a pH value of at least about 6.5, and in fact teaches away from such a formulation.

The preparation of the benzimidazole-containing compositions of the present invention is described first with reference to the following general description and then with reference to the following non-limiting examples of the preparation and application of the compositions of the present invention.

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As noted previously, the formulation of the present invention includes a substrate which features the benzimidazole derivative. A solution is prepared with the enteric coating material, which has a pH value of at least 6.5 and more preferably of from about 7 to about 10. Preferably, a pH value in the desired range is obtained by adding an alkaline compound to an enteric coating material. More preferably, the alkaline compound is selected from the group consisting of sodium, potassium or ammonium hydroxide. This enteric coating solution is then layered directly over the substrate to form the composition of the present invention.

The term "substrate" refers to substantially any structure which features the benzimidazole derivative, such as Omeprazole. For example, this structure could be an active core containing the benzimidazole derivative. This active core could be prepared in a number of different ways which are known in the art. For example, the active core could be formed by compressing the benzimidazole derivative with an alkaline substance. As another example, the active core could be prepared by mixing the benzimidazole derivative with an alkaline substance, spheronizing the mixture and then forming cores through pelletisation. As yet another example, the active core is optionally and preferably prepared by embedding the active ingredient in a poloxamer and compressing the embedded material into tablets. The active core is also optionally formed by granulating the active ingredient with an alkaline substance and compressing the granulation into tablets.

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Alternatively and optionally, the structure could include a neutral core, such as a sugar bead which does not contain the benzimidazole derivative, over which the benzimidazole derivative is coated. The coating includes Omeprazole or other benzimidazole derivative with a suitable adhesive polymer.

The substrate optionally and preferably includes a basic stabilizing material, which is more preferably at least one of sodium stereate and arginine, particularly for the active coating. Magnesium carbonate and/or sodium hydrogen carbonate may also optionally be used as basic stabilizing materials, in addition to, or alternatively in place of, these materials.

Substantially any type of neutralized suitable enteric coating material could be used in order to coat the benzimidazole substrate, including but not limited to, cellulose

acetate phthalate (CAP); hydroxypropyl methylcellulose phthalate (HPMCP); polyvinyl acetate phthalate; cellulose acetate trimellitate; polymethacrylic acid methyl methacrylate or ethyl methacrylate, such as the various types of Eudragit; and hydroxypropyl methylcellulose acetate succinate (HPMCAS). However, preferably the enteric coating material is prepared with the proviso that this material does not contain HPMCP alone, but only in combination with at least one of these other listed enteric coating materials. More preferably, HPMCP is not present in the enteric coating material. The particularly preferred enteric coating material is HPMCAS.

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As used herein, the term "neutralized enteric coating material" refers to enteric coating material which has been at least partially neutralized by reaction with an alkaline compound, which is optionally a basic inorganic salt. Preferably, the enteric coating material is at least about 60 % neutralized, more preferably the enteric coating material is at least about 80 % neutralized, and most preferably the enteric coating material is at least about 95 % neutralized.

The enteric coating optionally contains a plasticizer, such as a citric acid ester, a phthalic acid ester, or any suitable plasticizer.

The method for applying the enteric coating material to the substrate can vary. Substantially any coating method can be used, such as pan coating or fluidized bed coating, with the solution of the enteric coat chosen. As noted previously, preferably this solution is an aqueous solution. The enteric coating materials described previously can be applied to the substrate in an aqueous solution if the pH value of the solution is adjusted to at least 6.5, and more preferably to an alkaline value, most preferably a pH value from about 7 to about 10.

The following specific examples illustrate various aspects of the compositions of the present invention, and are not intended to be limiting in any way. Specific reference is made to Omeprazole for the purposes of description only and without intending to be limiting.

Example 1

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This example of the composition of the present invention was prepared as follows. The substrate was in the form of an active core, which was prepared by embedding Omeprazole in poloxamer (Pluronic PE 6800), granulating the resulting mass, adding the necessary auxiliary substances to the mass, and compressing the resultant material into tablets. The substrate was then coated with alkaline polyvinyl acetate phthalate as the enteric coating layer.

Substrate (Active Embedded Core)

<u>Ingredients</u>	Quantity per tablet
Omeprazole	20 mg
Poloxamer (Pluronic PE 6800)	200 mg
Colloidal silicon dioxide	7 mg
Magnesium carbonate	10 mg
Sodium starch glycolate	12 mg
Titanium dioxide	100 mg
Ludipress ®	226 mg
Sodium stearyl fumarate	25 mg

Enteric coating layer

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Polyvinyl acetate phthalate	75 mg
Antifoam emulsion	0.25 mg
Sodium hydroxide	12 mg

For the preparation of the substrate, the poloxamer was melted at a temperature of 80 °C. Omeprazole, together with 2 mg colloidal silicon dioxide, 8 mg of magnesium carbonate, titanium dioxide and 6 mg of sodium starch glycolate were added and mixed thoroughly. Mixing was continued until the melt solidified. The melt was granulated and the rest of the ingredients added to the granulate. The granulate was then compressed into tablets which contained 20 mg Omeprazole. These tablets, which formed the substrate of the composition, were then transferred into a conventional coating pan and coated with

the enteric coating layer, prepared in the following manner. First, the antifoam emulsion was dissolved in water to form an aqueous solution. Polyvinyl acetate phthalate was then stirred into this solution for a final concentration of about 10% weight per volume before sodium hydroxide was added. Sodium hydroxide (1 M solution) was then added to adjust the pH value of the solution to about 8, thereby obtaining a basic solution of the enteric coating material. This solution was then sprayed onto the tablets with an incoming air temperature of 40 °C.

Example 2

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This example of the composition of the present invention was prepared as follows. The substrate was prepared by embedding Omeprazole in poloxamer (Pluronic PE 6800) to form tablets, as for Example 1. However, in this Example, the tablets were then coated with hydroxypropyl methylcellulose acetate succinate (HPMCAS) as the enteric coating layer.

Substrate

<u>Ingredients</u>	Quantity per tablet	
Omeprazole	20 mg	
Poloxamer (Pluronic PE 6800)	200 mg	
Colloidal silicon dioxide	7 mg	
Sodium starch glycolate	20 mg	
Ludipress ®	228 mg	
Sodium stearyl fumarate	25 mg	

Enteric coating layer

Hydroxypropyl Methylcellulose Acetate	43 mg
Succinate	
(HPMCAS)	
Triethyl citrate	12 mg
Sodium lauryl sulfate	1.3 mg
Talc	21.4 mg
Sodium hydroxide	2.3 mg

The tablets were prepared as for Example 1, except that titanium dioxide was omitted. The tablets were then coated in a conventional coating pan with the enteric coating solution, which was prepared as follows. First, triethyl citrate was dissolved in water to form an aqueous solution. Sodium lauryl sulfate was then added to this aqueous solution. The HPMCAS and talc were dispersed in this solution, such that the concentration of HPMCAS was about 10% weight per volume. Sodium hydroxide (1M solution) was then added to adjust the pH value of the solution to a value from about 7 to about 10. The enteric coating was layered over the substrate by spraying the solution with an incoming air temperature of 40 °C.

Example 3

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This example of the composition of the present invention was prepared as for

Example 1, except that the enteric coating contained alkaline HPMCP

(hydroxypropylmethylcellulose phthalate) rather than HPMCAS.

Substrate

<u>Ingredients</u>	Quantity per tablet
Omeprazole	20 mg
Poloxamer (Pluronic PE 6800)	200 mg
Colloidal silicon dioxide	7 mg
Sodium starch glycolate	10 mg
Titanium dioxide	83 mg
Ludipress ®	145 mg
Sodium stearyl fumarate	25 mg

Enteric coating layer

HPMC Phthalate (HP-55)	56.2 mg
Triethyl citrate	22.5 mg
Sodium hydroxide	9 mg

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The substrate was prepared as described in Example 1, and was then coated in a conventional coating pan with the enteric coating solution by spraying the solution at an incoming air temperature of 40 °C. The enteric coating solution was prepared as follows. The HPMC phthalate was suspended in the water to a concentration of about 10% weight per volume (before sodium hydroxide was added). Sodium hydroxide (1M solution) was then added to this aqueous suspension until the HPMCP dissolved. The resultant solution has a pH value in a range of from about 8 to about 10. The triethyl citrate was then added

to the resultant solution in order to form the enteric coating solution, which was then layered over the substrate as previously described.

Example 4

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In this example of the composition of the present invention, the substrate has two parts: a neutral core; and a coating layer containing the active ingredient, which was layered over the neutral core. The substrate was then coated with the enteric coating solution. Hard gelatin capsules were then filled with the resultant pellets.

Substrate

Neutral core

Quantity per capsule

Sugar spheres 20/25 (700-850 microns)	161.63 mg

Active coating

<u>Ingredients</u>	Quantity per capsule
Omeprazole	20.00 mg
Hydroxypropyl methylcellulose 2910	5.33 mg
Hydroxypropyl cellulose	6.00 mg
Lactose	8.00 mg
Disodium phosphate anhydrous	0.64 mg
Sodium lauryl sulfate	0.50 mg

Enteric coating layer

HPMCAS	21.00 mg
Triethyl citrate	6.00 mg
Sodium lauryl sulfate	0.66 mg
Talc	11.00 mg
Sodium hydroxide	1.12 mg

The composition of the present invention was prepared according to this Example as follows. First, sugar spheres were placed in a fluid bed coating chamber, equipped with a Wurster bottom spraying device. A suspension of the ingredients in water was then prepared so that the concentration was approximately 20 % of total solids in water. This active coating suspension was sprayed onto the sugar spheres. A suspension of the enteric coating was prepared according to Example 2. This enteric coating was then sprayed onto the substrate in order to form the finished pellets. The pellets were then placed in capsules.

Example 5

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This example of the composition of the present invention was prepared with a compressed tablet as the substrate. The tablet was then coated with alkaline HPMCAS (Hydroxypropyl Methylcellulose Acetate Succinate) as the enteric coating layer, preferably having a pH in a range of from about 7 to about 10.

Substrate (Active Compressed Tablet Core)

<u>Ingredients</u>	Quantity per tablet	
Omeprazole	20 mg	
Lactose	192.5 mg	
Magnesium carbonate	10 mg	
Sodium starch glycolate	10 mg	
Povidone	10 mg	
Sodium stearyl fumarate	7.5 mg	

Enteric coating layer

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HPMCAS	16.1 mg	
Triethyl citrate	4.5 mg	
Sodium lauryl sulfate	0.5 mg	
Talc	8.04	
Sodium hydroxide	0.86 mg	

For the preparation of the substrate, Omeprazole, together with lactose, magnesium carbonate, sodium starch glycolate, and povidone were mixed thoroughly. The mixture was then granulated with a sufficient quantity of water, and dried. Sodium stearyl fumarate was then added to the mixture, which was then compressed into tablets weighing 250 mg each.

These tablets, which formed the substrate of the composition, were then transferred into a conventional coating pan and coated with the enteric coating layer, prepared as described in Example 4.

5 Example 6

Stability tests were performed with formulations prepared according to Examples 2 and 3. For the first test, both coated and uncoated tablets prepared according to either Example 2 or Example 3 were placed into a box which was open to the environment. The open box was then stored at 40 °C and 75 % relative humidity, which are very stringent conditions. The coated and uncoated tablets were examined initially, after a week and after a month to determine stability. The results are shown in the tables below.

Tablets Prepared According to Example 2

Sampled Material	Appearance of Sample		
	<u>Initial</u>	After One Week	After One Month
coated tablet	off white	deeper off white	deeper off white
uncoated tablet	white	white	white

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<u>Tablets Prepared According to Example 3</u>

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Sampled Material	Appearance of Sample		
	<u>Initial</u>	After One Week	After One Month
coated tablet	off white	off white	deeper off white
uncoated tablet	white	white	white

The term "deeper off white" refers to a more intense off white color which was observed for some samples, as described in greater detail above. These results show that coated tablets prepared according to either Example 2 or Example 3 showed good stability, even after one month of storage under particularly stringent conditions.

In a second stability test, coated tablets were prepared according to Example 2. These coated tablets were then packed into an Alu/Alu (Aluminum/Aluminum) blister, which is a well known technique in the art for packing certain oral dosage forms. The blister was then stored under accelerated conditions of 30 °C and 60% relative humidity; or 40 °C and 75% relative humidity. Samples of the tablets were examined initially, and after one month of storage under one of these conditions. In addition, samples were assayed to determine the amount of Omeprazole present in the coated tablet, as listed under "Assay" as milligrams of Omeprazole per tablet. A dissolution test was performed, using the accepted USP method. The coated tablets were placed in 0.1 N HCl for 2 hours, followed by a solution at pH 6.8 with stirring with a paddle at 100 rpm for 15 minutes, 30 minutes or 45 minutes. Gastric resistance was also examined by placing the coated tablets in a simulated gastric fluid for 2 hours (pH of approximately 1), as is well known in the art. The results are shown in the table below.

	Time (min)	<u>Initial</u>	30° 60 % RH	40 ⁰ 75% RH
Description	NA	Off white	Off white	Off white
Assay	NA	20.4 mg	19.39 mg	19.66 mg
Dissolution	120	0%	0%	0%
	135	52%	42%	39%
	150	96%	85%	90%
	165	105%	99%	104%
Gastric	NA	101%	98%	96%
Resistance				

These results show that the coated tablets, prepared according to Example 2, show good stability and gastric resistance, yet are also able to dissolve in an appropriate time-dependent manner.

Example 7

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A one-way pharmacokinetic pilot study was performed *in vivo* for testing the pharmacokinetic profile of the coated tablets, which were prepared according to Example 2. The study was performed with ten healthy male volunteers, who received a single dosage of the coated tablets, containing 20 mg of Omeprazole. The results showed that Omeprazole administered in the coated tablets of the present invention had a similar lag time to absorption in comparison to a previous study performed with the reference product, which is the 20 mg Omeprazole dosage form of the formulation of Astra (Aktiebolaget Hassle),

and also as described in the literature (see for example Duvauchelle, T. et al., "Comparative Bioavailability Study of Two Oral Omeprazole Formulations After Single and Repeated Administrations in Healthy Volunteers", *Pharmacokinetics*, **16**: 141-149, 1998). The lag time to absorption is defined as the time between the administration of the formulation and the first detection of the active ingredient in the samples taken from the subject, according to the sampling method employed.

In addition, comparable bioavailability was achieved with the coated tablets of the present invention, both to values obtained in the previous study with the reference product and to values which were described in the literature (see for example the previously referenced article in *Pharmacokinetics*). Furthermore, the values obtained for Cmax and Tmax concerning the rate of absorption were comparable to results obtained in the previous study performed with the reference product, and as described in the literature (see for example the previously referenced article in *Pharmacokinetics*). Thus, the coated tablets of the present invention clearly show good performance both *in vitro*, as described in Example 6, and *in vivo*.

Example 8

Coated pellets were prepared according to the process previously described above in Example 4. However, the pellets were coated with the following suspension:

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Enteric coating (quantities per capsule)

HPMCAS	21.00 mg
Triethyl Citrate	6.00 mg
Sodium lauryl sulfate	0.66 mg
Colloidal silicon dioxide	2.10 mg
Sodium hydroxide	1.12 mg

Example 9

Although the previous Examples used aqueous solutions for providing an optimal coating, the possibility of increasing the concentration of the enteric coating polymer by using an alcohol-based solution was studied in this Example.

Coated pellets were prepared according to the process of Example 4, except that these pellets were coated with the following solution, to obtain the required protection in an acidic environment.

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Enteric coating

	Solution prepared	Quantities per capsule
Alcohol 95%	1.900 kg	N/A
Water	0.830 kg	N/A
HPMCAS	0.476 kg	21.00 mg
Triethyl citrate	0.136 kg	6.00 mg
Sodium lauryl sulfate	0.015 kg	0.66 mg
Colloidal silicon dioxide	0.047 kg	2.1 mg
Sodium hydroxide	0.025 kg	1.12 mg

Example 10

<u>Substrate (Active Compressed Tablet Core)</u>

<u>Ingredients</u>	Quantity per tablet
Omeprazole	20mg
Lactose	203mg
Magnesium carbonate	10mg
Sodium starch glycolate	10mg
Sodium stearyl fumarate	7mg

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<u>Ingredients</u>	Quantity per tablet
HPMCAS	16mg
Triethyl citrate	4.5mg
Sodium lauryl sulfate	0.5mg
Talc	8.14mg
Sodium hydroxide	0.86mg
Sepisperse [™] (pink pigment)	10.8mg

For the preparation of the substrate, Omeprazole was mixed together thoroughly with lactose, sodium starch glycolate, magnesium carbonate and sodium stearyl fumarate. The mixture was then compressed into tablets weighing 250mg each. These tablets were then

transferred into a conventional coating pan and coated with the enteric coating layer, prepared as described in Example 4, with the addition of a pigment to the enteric coating material.

5 Example 11

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Stability tests were performed with the formulation prepared according to Example 10. For the tests, the tablets were packed into alu-alu blister. The blister was then stored under room temperature or under accelerated conditions of 30°C and 60% relative humidity (RH), or 40°C and 75% relative humidity. Samples of the tablets were examined initially and after 6 months of storage under one of these conditions. In addition samples were assayed. A dissolution test was performed, and gastric resistance was also examined. The tablet gave good stability results even after storage at 40°C. The results are shown in the table below.

Test	Initial	25°C	30°C /	40°C /
performed		6 month	60%RH	75%RH
			6 month	6 month
Visual	conform	conform	conform	conform
Description				
Assay	19.76mg per	20.19mg per	19.97mg per	19.28mg per
	tablet	tablet	tablet	tablet
Dissolution	96%	96%	96%	96%
Gastric	96%	96%	95%	94%
Resistance				

Example 12

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A two-way pharmacokinetic study was performed in vivo for testing the bioequivalence of the coated tablets which were prepared according to Example 10, as compared to the reference product which is the 20mg Omeprazole dosage form of the formulation of Astra (Sweden), called LosecTM. The study was performed on 39 volunteers. As shown in the table below, the results of the study showed that the two products exhibited very similar pharmacokinetic profiles, such that the two formulations

can be considered to be bioequivalent.

Formulation	AUC	Cmax	Tmax
	(ng x hour/ml)	(ng/ml)	(hours)
Formulation of the	426 ± 256	217 ± 109	1.08 ± 0.64
present invention			
(Example 10)			
Losec TM (Astra)	434 ± 226	246 ± 113	1.56 ± 0.79

Example 13

5 <u>Substrate (Active Compressed Tablet Core)</u>

<u>Ingredients</u>	Quantity per tablet
Omeprazole	20mg
Lactose	203mg
Sodium hydrogen carbonate	10mg
Sodium starch glycolate	10mg
Sodium stearyl fumarate	7mg

<u>Ingredients</u>	Quantity per tablet
HPMCAS	16mg
Triethyl citrate	4.5mg
Sodium lauryl sulfate	0.5mg
Talc	8.14mg
Sodium hydroxide	0.86mg
Sepisperse TM	10.8mg

For the preparation of the substrate, Omeprazole was thoroughly mixed together with lactose, sodium starch glycolate, sodium hydrogen carbonate and sodium stearyl fumarate. The mixture was then compressed into tablets weighing 250mg each. These tablets were then transferred into a conventional coating pan and coated with the enteric coating layer, prepared as described in Example 4.

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Example 14

Substrate (Active Compressed Tablet Core)

<u>Ingredients</u>	Quantity per tablet
Omeprazole	20mg
Lactose	203mg
Trisodium citrate	10mg
Sodium starch glycolate	10mg
Sodium stearyl fumarate	7mg

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<u>Ingredients</u>	Quantity per tablet
HPMCAS	16mg
Triethyl citrate	4.5mg
Sodium lauryl sulfate	0.5mg
Talc	8.14mg
Sodium hydroxide	0.86mg
Sepisperse TM	10.8mg

For the preparation of the substrate, Omeprazole was mixed thoroughly together with lactose, sodium starch glycolate, trisodium citrate and sodium stearyl fumarate. The mixture was then compressed into tablets weighing 250mg each. These tablets were then

transferred into a conventional coating pan and coated with the enteric coating layer, prepared as described in Example 4.

Example 15

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Stability tests were performed with the formulations prepared according to Examples 10, 13 and 14. Both coated and non-coated tablets were placed into an open box and stored at 40°C and 75% relative humidity, which are very stringent conditions. The coated and uncoated tablets were examined initially after 1 week and again after 2 weeks to determine stability. The results are shown in the tables below.

Tablets prepared according to Example 10

Sampled material	Appearance of sample		ole
	Initial	After 1 week	After 2 weeks
Coated	Pink	Pink	Pink
Uncoated	White	White	White

Tablets prepared according to Example 13

Sampled material	Appearance of sample		
	Initial	After 1 week	After 2 weeks
Coated	Pink	Pink	Pink
Uncoated	White	White	White

Tablets prepared according to Example 14

Sampled material	Appearance of sample		
	Initial	After 1 week	After 2 weeks
Coated	Pink	Pink	Pink
Uncoated	White	White	White

Example 16

In this example of the composition of the present invention, the substrate has two
parts: a neutral core; and a coating layer containing the active ingredient, which was
layered over the neutral core. The substrate was then coated with the enteric coating
solution. Hard gelatin capsules were then filled with the resultant pellets.

Substrate

10 Neutral core

Quantity per capsule

Sugar spheres 20/25 (700-850 microns)	110 mg

Active coating

<u>Ingredients</u>	Quantity per capsule
Omeprazole	20.00 mg
Hydroxypropyl methylcellulose 2910	5.00 mg
Hydroxypropyl cellulose	6.00 mg
Sodium stearate	0.13 mg
Sodium lauryl sulfate	0.50 mg

Enteric coating layer

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HPMCAS	21.00 mg
Triethyl citrate	6.00 mg
Sodium lauryl sulfate	0.66 mg
colloidal silicon dioxide	2.1 mg
Arginine	3.15 mg

The composition of the present invention was prepared according to this Example as follows. First, sugar spheres were placed in a fluid bed coating chamber, equipped with a Wurster bottom spraying device. A suspension of the ingredients in water was then prepared so that the concentration was approximately 20% of the total solids in water. This active coating suspension was sprayed onto the neutral sugar spheres.

A suspension of the enteric coating was prepared as follows. First, triethyl citrate was dissolved in water to form an aqueous solution. Sodium lauryl sulfate was then added to this aqueous solution. HPMCAS and colloidal silicon dioxide were dispersed in

this solution, such that the concentration of HPMCAS was about 10% weight per volume. Arginine (3% weight per volume solution) was added to adjust the pH value of the solution to a pH value in a range of from about pH 7 to about pH 9. The enteric coating was layered over the substrate in order to form the finished pellets. The pellets were then placed in capsules.

Example 17

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In this example of the composition of the present invention, the substrate has two parts: a neutral core; and a coating layer containing the active ingredient, which was layered over the neutral core. The substrate was then coated with the enteric coating solution. Hard gelatin capsules were then filled with the resultant pellets.

Substrate

Neutral core

Quantity per capsule

Sugar spheres 20/25 (700-850 microns)	110 mg

Active coating

<u>Ingredients</u>	Quantity per capsule
Omeprazole	20.00 mg
Hydroxypropyl methylcellulose 2910	5.00 mg
Hydroxypropyl cellulose	6.00 mg
Arginine	0.13 mg
Sodium lauryl sulfate	0.50 mg

Enteric coating layer

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HPMCAS	21.00 mg
Triethyl citrate	6.00 mg
Sodium lauryl sulfate	0.66 mg
colloidal silicon dioxide	2.1 mg
Sodium hydroxide	1.12 mg

The composition of the present invention was prepared according to this Example as follows. First, sugar spheres were placed in a fluid bed coating chamber, equipped with a Wurster bottom spraying device. A suspension of the ingredients in water was then prepared so that the concentration was approximately 20% of the total solids in water. This active coating suspension was sprayed onto the sugar spheres.

A suspension of the enteric coating was prepared according to Example 8. The enteric coating was layered over the substrate in order to form to form the finished pellets. The pellets were then placed in capsules.

Example 18

In this example of the composition of the present invention, the substrate has two parts: a neutral core; and a coating layer containing the active ingredient, which was layered over the neutral core. The substrate was then coated with the enteric coating solution. Hard gelatin capsules were then filled with the resultant pellets.

Substrate

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Neutral core

Quantity per capsule

Sugar spheres 20/25 (700-850 microns)	110 mg

10 Active coating

<u>Ingredients</u>	Quantity per capsule
Omeprazole	20.00 mg
Hydroxypropyl methylcellulose 2910	5.00 mg
Hydroxypropyl cellulose	6.00 mg
Arginine	0.13 mg
Sodium lauryl sulfate	0.50 mg

HPMCAS	21.00 mg
Triethyl citrate	6.00 mg
Sodium lauryl sulfate	0.66 mg
colloidal silicon dioxide	2.1 mg
Arginine	3.15 mg

The composition of the present invention was prepared according to this Example as follows. First, sugar spheres were placed in a fluid bed coating chamber, equipped with a Wurster bottom spraying device. A suspension of the ingredients in water was then prepared so that the concentration was approximately 20% of the total solids in water. This active coating suspension was sprayed onto the sugar spheres to form the substrate.

A suspension of the enteric coating was prepared according to Example 16. The enteric coating was layered over the substrate in order to form to form the finished pellets. The pellets were then placed in capsules.

Example 19

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In this example of the composition of the present invention, the substrate has two parts: a neutral core; and a coating layer containing the active ingredient, which was layered over the neutral core. The substrate was then coated with the enteric coating solution to form pellets. Hard gelatin capsules were then filled with the resultant pellets.

Substrate

Neutral core

<u>Ingredients</u>	Quantity per capsule
Sugar spheres 20/25 (700-800 microns)	110mg

Active coating

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<u>Ingredients</u>	Quantity per capsule
Omeprazole	20 mg
Hydroxypropyl methylcellulose 2910	5 mg
Hydroxypropyl cellulose	6 mg
Sodium lauryl sulfate	0.5 mg
Arginine	0.1 mg

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<u>Ingredients</u>	Quantity per capsule
Triethyl citrate	35 mg
Sodium lauryl sulfate	3.8 mg
HPMCAS	126 mg
Colloidal silicon dioxide	19 mg
Talc	17 mg
Ammonia (in a 25% solution)	3 mg

The composition of the present invention was prepared according to this example as follows. First, sugar spheres were placed in a fluid bed-coating chamber, equipped with a Wurster bottom-spraying device. Next, a suspension of the ingredients in water was then prepared for a final concentration of the total solids of approximately 15% in water, to form the active coating. This active coating suspension was sprayed onto the sugar spheres, thereby forming the substrate.

A suspension of the enteric coating was then prepared as follows. First, triethyl citrate was dissolved in water to form an aqueous solution. Sodium lauryl sulfate was then added to this aqueous solution. HPMCAS, colloidal silicon dioxide and talc were dispersed in this solution, such that the concentration of HPMCAS was about 7% weight per volume. Ammonia in a 25% solution was added to adjust the pH value in a range of from about 7 to about pH 9. The enteric coating was layered over the substrate in order to form the finished pellets. The pellets were then placed in capsules.

Example 20

Substrate (active compressed tablet core)

<u>Ingredients</u>	Quantity per capsule
Omeprazole	20 mg
lactose	203 mg
Magnesium carbonate	10 mg
Sodium starch glycollate	10 mg
Sodium stearyl fumarate	7 mg

Enteric coating layer

<u>Ingredients</u>	Quantity per capsule	
Triethyl citrate	4.5 mg	
Sodium lauryl sulfate	0.5 mg	
HPMCAS	16 mg	
Talc	8.14 mg	
Ammonia (in a 25% solution)	0.14 mg	
Sepisperse ® (pink pigment)	10.8 mg	
Isopropyl alcohol	N/A	
Alcohol	N/A	

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The substrate of the prevent invention was prepared as described in Example 10. A suspension of the enteric coating was then prepared as follows. First, triethyl citrate was dissolved in a mixture of isopropyl alcohol and alcohol. Sodium lauryl sulfate was then

added to this solution. HPMCAS and talc were dispersed in this solution, such that the concentration of HPMCAS was about 6% weight per volume. Ammonia in a 25% solution was added to adjust the pH value in a range of from about pH 7 to about pH 9. The pigment was then added to the enteric coating dispersion. The tablet cores were then transferred into a conventional coating pan and coated with the enteric coating layer.

Example 21
Substrate (active compressed tablet core)

<u>Ingredients</u>	Quantity per capsule			
Omeprazole	10 mg			
lactose	101.5 mg			
Sodium stearate	5 mg			
Sodium starch glycollate	5 mg			
Sodium stearyl fumarate	3.5 mg			

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<u>Ingredients</u>	Quantity per capsule			
Triethyl citrate	2.25 mg			
Sodium lauryl sulfate	0.25 mg			
HPMCAS	8 mg			
Talc	4.7 mg			
Sodium hydroxide	0.43 mg			
Sepisperse ® (pink pigment)	5.4 mg			

For the preparation of the substrate, Omeprazole was mixed together thoroughly with lactose, sodium starch glycollate, sodium stearate and sodium stearyl fumarate. The mixture was then compressed into tablets weighing 125 mg each. These tablet were then transferred into a conventional coating pan and coated with the enteric coating layer, prepared as described in Example 10.

Example 22

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Stability tests were performed with the formulation prepared according to Example 21. For the tests, the tablets were packed into alu-alu blister. The blister was then stored under room temperature or under accelerated conditions of 30°C and 60% relative humidity (RH), or 40°C and 75% relative humidity. Samples of these tablets were examined initially and after 6 months of storage under one of these conditions. In addition samples were assayed and purity test was performed. A dissolution test was

performed, and gastric resistance was also examined. The tablet gave good stability results even after storage at 40°C. The results are shown in the table below.

Test performed	Initial	25°C	30°C/60%RH	40°C/75%RH
		6 months	6 months	6 months
Description	conform	conform	conform	conform
Assay	98.4%	96.7%	96.9%	96.3%
Dissolution	95%	97%	95%	95%
gastric resistance	96%	95%	97%	96%
individual impurity	0.04%	not	not detectable	0.23%
total impurity	0.04%	detectable	not detectable	0.29%
		not		
		detectable		

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While the invention has been described with respect to a limited number of embodiments, it will be appreciated that many variations, modifications and other applications of the invention may be made.

WHAT IS CLAIMED IS:

- 1. A stable composition for a benzimidazole derivative, the composition comprising:
 - (a) a substrate, said substrate featuring the benzimidazole derivative; and
 - (b) an enteric coating material layered directly over said substrate, said enteric coating material having a pH value of at least about 6.5, thereby obviating the need for an intermediate layer between said substrate and said enteric coating, with the proviso that said enteric coating material does not include HPMCP (hydroxypropyl methylcellulose phthalate).
- 2. The composition of claim 1, wherein said substrate is an active core for containing the benzimidazole derivative.
- 3. The composition of claim 2, wherein said active core is selected from the group consisting of a pellet, a bead and a tablet.
- 4. The composition of claim 2, wherein said active core is a tablet formed by compression.
 - 5. The composition of claim 1, wherein said substrate features:
 - (i) a neutral core; and .

- (ii) an active coating containing the benzimidazole derivative, said active coating being layered over said neutral core; such that the composition is in a form of a pellet.
- 6. The composition of claim 5, wherein said active coating includes at least one polymer.
- 7. The composition of claim 6, wherein said at least one polymer is selected from the group consisting of hydroxypropyl methylcellulose (HPMC) and hydroxypropyl cellulose (HPC).
 - 8. The composition of claim 6, wherein said active coating includes a combination of hydroxypropyl methylcellulose (HPMC) and hydroxypropyl cellulose (HPC).
 - 9. The composition of claim 6, wherein said active coating includes a basic stabilizing material.
 - 10. The composition of claim 9, wherein said basic stabilizing material includes at least one of sodium stereate and arginine.
- 11. The composition of claim 1, wherein said substrate includes a basic stabilizing material.

- 12. The composition of claim 11, wherein said basic stabilizing material includes at least one of sodium stereate, arginine, magnesium carbonate and sodium hydrogen carbonate.
- 13. The composition of claim 1, wherein said substrate features a core containing the benzimidazole derivative with a suitable binding agent, said core being prepared by spheronisation and pelletization; such that the composition is in a form of a pellet.
- 14. The composition of claim 1, wherein said enteric coating material includes at least one enteric material selected from the group consisting of hydroxypropyl methylcellulose acetate succinate, polyvinyl acetate phthalate, cellulose acetate phthalate, cellulose acetate trimellitate, polymethacrylic acid methyl methacrylate and polymethacrylic acid ethyl methacrylate.
- 15. The composition of claim 14, wherein said enteric coating material further comprises an alkaline compound, such that said pH value is adjusted by adding said alkaline compound to said enteric material.
- 16. The composition of claim 15, wherein said alkaline compound is an inorganic alkaline compound.

- 17. The composition of claim 16, wherein said alkaline compound is selected from the group consisting of basic sodium, potassium and ammonium hydroxide.
- 18. The composition of claim 17, wherein said enteric coating material is at least about 60 % neutralized by adding said alkaline compound.
- 19. The composition of claim 18, wherein said enteric coating material is at least about 80 % neutralized by adding said alkaline compound.
- 20. The composition of claim 18, wherein said enteric coating material is at least about 95 % neutralized by adding said alkaline compound.
- 21. The composition of claim 15, wherein said pH value is in a range of from about 7 to about 10.
- 22. The composition of claim 15, wherein said enteric coating material further comprises a plasticizer.
- 23. The composition of claim 22, wherein said plasticizer is selected from the group consisting of a citric acid ester and a phthalic acid ester.

- 24. The composition of claim 1, wherein the benzimidazole derivative is selected from the group consisting of Omeprazole, Pantoprazole, Lansoprazole, Leminoprazole, Perprazole, Rabeprazole, and pharmaceutically acceptable salts thereof.
- 25. A stable composition for a benzimidazole derivative, the composition consisting essentially of:
 - (a) a substrate, said substrate featuring the benzimidazole derivative; and
 - (b) an enteric coating material layered over said substrate, said enteric coating material having a pH value of at least about 6.5 by an alkaline compound, such that said pH value is adjusted by adding said alkaline compound to said enteric material.
- 26. The composition of claim 25, wherein said substrate is an active core for containing the benzimidazole derivative.
- 27. The composition of claim 26, wherein said active core is selected from the group consisting of a pellet, a bead and a tablet, said active core being formed by embedding the benzimidazole derivative in poloxamer.
- 28. The composition of claim 26, wherein said active core is a tablet formed by compression.
 - 29. The composition of claim 25, wherein said substrate features:

- (i) a neutral core; and
- (ii) an active coating containing the benzimidazole derivative, said active coating being layered over said neutral core.
- 30. The composition of claim 29, wherein said active coating includes at least one polymer.
- 31. The composition of claim 30, wherein said at least one polymer is selected from the group consisting of hydroxypropyl methylcellulose (HPMC) and hydroxypropyl cellulose (HPC).
- 32. The composition of claim 30, wherein said active coating includes a combination of hydroxypropyl methylcellulose (HPMC) and hydroxypropyl cellulose. (HPC).
- 33. The composition of claim 29, wherein said enteric coating material includes at least one enteric material selected from the group consisting of hydroxypropyl methylcellulose phthalate, hydroxypropyl methylcellulose acetate succinate, polyvinyl acetate phthalate, cellulose acetate phthalate, cellulose acetate trimellitate, polymethacrylic acid methyl methacrylate and polymethacrylic acid ethyl methacrylate.

- 34. The composition of claim 33, wherein said alkaline compound is an inorganic alkaline salt compound.
- 35. The composition of claim 34, wherein said alkaline compound is selected from the group consisting of basic sodium, potassium or ammonium hydroxide.
- 36. The composition of claim 35, wherein said enteric coating material is at least about 60 % neutralized by adding said alkaline compound.
- 37. The composition of claim 36, wherein said enteric coating material is at least about 80 % neutralized by adding said alkaline compound.
- 38. The composition of claim 37, wherein said enteric coating material is at least about 95 % neutralized by adding said alkaline compound.
- 39. The composition of claim 34, wherein said pH value is in a range of from about 7 to about 10.
- 40. The composition of claim 34, wherein said enteric coating material further comprises a plasticizer.
- 41. The composition of claim 40, wherein said plasticizer is selected from the group consisting of a citric acid ester and a phthalic acid ester.

- 42. The composition of claim 25, wherein the benzimidazole derivative is selected from the group consisting of Omeprazole, Pantoprazole, Lansoprazole, Leminoprazole, Perprazole, Rabeprazole, and pharmaceutically acceptable salts thereof.
- 43. A method for producing a stable composition for a benzimidazole derivative, the method comprising:

forming a substrate with the benzimidazole derivative;

preparing an enteric coating material having a pH value of at least about 6.5; and layering said enteric coating material directly over said substrate, with the proviso that said enteric coating material does not include HPMCP (hydroxypropyl methylcellulose phthalate).

- 44. The method of claim 43, wherein said substrate is formed by melting poloxamer and by mixing the benzimidazole derivative into said poloxamer.
- 45. The method of claim 43, wherein said substrate is formed by direct compression.
- 46. The method of claim 43, wherein said substrate is formed by wet granulation.

- 47. The method of claim 43, wherein said substrate is formed by coating on an inert core.
- 48. The method of claim 43, wherein said enteric coating material is prepared by:

 mixing an enteric material with water to form a mixture; and

adding an alkaline compound to said mixture to form an aqueous solution having a pH value of from about 7 to about 10.

49. The method of claim 43, wherein said enteric coating material is prepared by:

mixing an enteric material with water and alcohol to form a mixture; and adding an alkaline compound to said mixture to form an aqueous solution having a pH value of from about 7 to about 10.

ABSTRACT OF THE DISCLOSURE

A stable composition with a benzimidazole derivative, such as Omeprazole, which does not contain a separating layer between the active compound and an enteric coating layer. Instead, the enteric coating layer is applied as a solution with a pH value of at least 6.5, and more preferably in a range of from about 7 to about 10, directly to the benzimidazole derivative substrate. This solution, with the optional addition of a plasticizer, can be directly coated onto the substrate without any necessity for an intermediate layer. Furthermore, in this pH range, the enteric coating is optionally applicable in an aqueous solution, thereby obviating the need for organic solvents for dissolving the enteric coating material. The resultant formulation maintains the stability of the benzimidazole derivative during storage and at the same time protects the product during passage through the acidic environment of the stomach. The problem of interaction between the enteric coat and the alkaline core is thus completely eliminated as the enteric coat at this stage is no longer acidic.